Chemometric Analysis of Moxifloxacin and Metronidazole in Binary Drug Combinations With Spectrophotometric Method

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Abstract

this study. sensitive and accurate spectrophotometric-chemometric methods were developed for metronidazole and moxifloxacin without any separation step in pharmaceutical tablets. The used chemometric methods are partial least squares regression (PLS) and principal component regression (PCR). PLS and PCR were successfully applied for chemometric analysis of metronidazole and moxifloxacin in synthetic mixtures and pharmaceutical tablets. A concentration set including binary mixtures of metronidazole and moxifloxacin formed to 24 different combinations were randomly prepared in 0.1 M HCl. The accuracy and precision of the developed method were validated by analyzing synthetic mixtures containing the examined drugs. As a result of the determination, high recoveries and low standard deviations were found. Absorbance and concentration values were used in Minitab and other chemometric programs to calculate estimated concentrations with PCR and PLS. The second step, in drug tablets (Avelox and Flagy), was calculated amounts for metronidazole and moxifloxacin.

Keywords: Metronidazole and moxifloxacin, PLS, PCR.

1. Introduction

Metronidazole; 2-methyl-5-nitroimidazol-1-etanol is used as an antiprotozoal, antiamebic and antibacterial drug [1]. But metronidazole in high concentration can not eradicate all bacteria from carious lesions [2]. Metronidazole was examined spectrophotometric method [3,4], chromatographic method [5,6] and voltametric method [7,8].

Moxifloxacin; 1-cyclopropyl-7-[2,8-diazobicyclo (4.3.0) nonane]-6-fluoro-8-methoxy-1.4-dihydro-4-

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Chemometric calibration methods are observed that it is the best techniques to determinate the amount of each component in the complex mixture. The most accepted chemometric methods in drug analysis are principal component regression (PCR) and partial least squares regression (PLS) [22]. A relationship to be established between matrices of chemical data is determinated at chemometric methods [23].

In this study, principal component regression (PCR) and partial least squares regression (PLS) were successfully performed to simultaneous determination of metronidazole and moxifloxacin in a commercial tablet formulation, tablets without any separation method. Mean recoveries (%) and standard deviation of principal component regression (PCR) and partial least squares regression (PLS) methods were calculated for the validation of the methods. The acquired results were statistically compared each other.

2. Experimental

Chemicals and Reagents : All materials used were analytical grade. 0.025~g~/~250~ml metronidazole 0.025~g~/~250~ml moxifloxacin stock solutions were made up with 0.1~M~HCl. Mixtures were prepared at various ratios. A training set and validation set containing the drugs in various proportions, 24~synthetic~mixtures~synthetic~mixtures~(for validation)~was~made.

Instruments and Software: A Shimadzu UV-1700 PharmaSpec Spectrophotometer connected to an IBM PS with UV Probe Software was used for all measurements and data processing. A pair of 1.0 cm quartz cuvettes were used for absorbance measurements.

Absorbance Measurements: Interval corresponding to the difference 0.5 between 200-500 nm were recorded for the active ingredients of the absorbance spectra. Calibration matrix and training verification sets contain two component mixtures that are optimized at different ratios. Spectra analysis of pure agents and PLS and PCR were used to calculate the concentrations.

Samples of 5-35 ppm (either alone or in combination) between the active substances were taken up into 25 ml volumetric flask volumes and completed with 0.1 M HCl. The mixtures were agitated for thorough mixing.

Procedure for real sample: For this purpose, the active substance samples were transferred to 25 ml flasks and mechanically mixed in 0.1 M HCl in a medium similar to the stomach environment.

Chemometric Method: Partial least squares regression (PLS-regression) is the most commonly used chemometric multivariate calibration method [24].

PLS is done using both experimental (or x) and concentration (or c) data simultaneously. Usually PLS is presented in the form of equations. There are a few ways to express them, the most suitable for our purpose being:

$$X = T.P. + E \tag{1}$$

$$c = T.q + f \tag{2}$$

Where X refers to the experimental measurements (e.g. spectra) and c is the concentration. There is a correlation with an installation for vector vector q. Matrix T is common to both equations. E is an error matrix and error to prevent the x vector c to block scores f orthogonal, but non-orthogonal (P) loads, generally are non-normalized [25].

The Minitab 17 program (İnova, Ankara, Turkey) was used for the analysis of all the concentration and absorbance data and to do the statistical calculations. Minitab is a statistical analysis software. In addition to statistical research, statistics can be used to learn [27].

3. Results and Discussion

Figure 1 shows the absorbance-wave length (nm) curves. The spectrum of metronidazole and moxifloxacin is in the range of 200-500 nm.

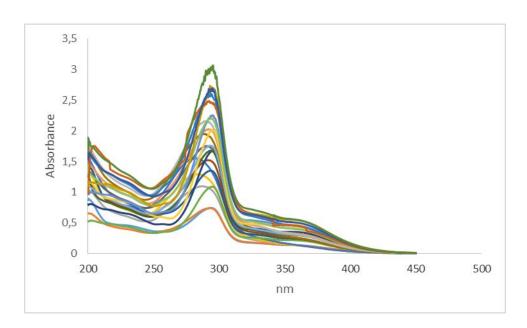


FIG. 1: The Absorption Spectrum of Metronidazole and Moxifloxacin

Fig.1 shows absorption spectra for metronidazole and moxifloxacin and their binary mixture in 0.1 M HCl.

Our objective in this study is to develop a lower-cost but more quick and reliable analytical method using chemometry. With this method, active ingredients can be analyzed without pre-separation, and loss of time and work due to the trial and error method will be prevented.

With the spectrophotometric-chemometric method used, the pharmaceutical industry will be faster and less costly. As a result, drug prices can be lowered so people can buy drugs more cheaply.

The PLS method and absorption spectra can be used individually or overlapping for multiple simultaneous detection of very linear components.

Concentrations of metronidazole and moxifloxacin mixtures are listed in Table 1. Prediction of the calibration methods was done by resolution of ten synthetic mixtures in a working concentration range for food colorants. Predicted concentration data of metronidazole and moxifloxacin mixtures is shown in Table 2. The percentage recovered and relative standard deviations are also listed in Table 2. The training set of 24 standard mixture solutions contain different concentrations of metronidazole and moxifloxacin.

Table 1: Concentration Set Containing Metronidazole and Moxifloxacin Compounds for The Preparation of The Pcr and Pls Calibration.

No	Moxifloxacin	Metronidazole	
	ppm	ppm	
1	5	7	
2	5	14	
3	5	21	
4	5	28	
5	5	35	
6	10	7	
7	10	14	
8	10	21	
9	10	28	
10	10	35	
11	15	7	
12	15	14	
13	15	21	
14	15	28	
15	15	35	
16	20	7	
17	20	14	
18	20	21	
19	20	28	
20	20	35	
21	25	7	
22	25	21	
23	25	28	
24	25	35	

Table 2: Composition of Prediction Set and Recovery Results Obtained in Synthetic Mixtures for Pls Method

	Moxifloxacin		Metronidazole			
Mixture No	Actual (ppm)	Predicted (ppm)	% Recovery	Actual (ppm)	Predicted (ppm)	% Recovery
1	5	4.85	97.00	7	6.98	99.71
2	5	4.97	99.40	14	13.92	99.43
3	5	4.94	98.80	21	20.95	99.76
4	5	4.86	97.20	28	27.89	99.61
5	5	5.01	100.20	35	34.79	99.40
6	10	9.86	98.60	7	6.97	99.57
7	10	9.87	98.70	14	13.94	99.57
8	10	9.97	99.70	21	19.96	95.05
9	10	10.00	100.00	28	27.93	99.75
10	10	9.97	99.70	35	34.97	99.91
11	15	14.96	99.73	7	6.92	98.86
12	15	14.95	99.67	14	13.85	98.93
13	15	14.97	99.80	21	20.97	99.86
14	15	14.93	99.53	28	27.99	99.96
15	15	14.98	99.87	35	34.82	99.49
16	20	19.85	99.25	7	6.89	98.43
17	20	19.87	99.35	14	13.96	99.71
18	20	19.82	99.10	21	20.98	99.90
19	20	19.89	99.45	28	27.89	99.61
20	20	19.97	99.85	35	34.96	99.89
21	25	24.78	99.12	7	6.98	99.71
22	25	24.98	99.92	21	20.95	99.76
23	25	24.97	99.88	28	27.97	99.89
24	25	25.01	100.04	35	34.96	99.89
			Mean =99.33 SS =0.808			Mean =99.40 SS =0.999

Table 3: Composition of prediction set and recovery results obtained in synthetic mixtures for Pcr method

	Moxifloxacin			Metronidazole		
Mixture No	Actual (ppm)	Predicted (ppm)	% Recovery	Actual (ppm)	Predicted (ppm)	% Recovery
1	5	4.98	99.60	7	7.01	100.14
2	5	4.88	97.60	14	13.85	98.93
3	5	4.92	98.40	21	20.86	99.33
4	5	4.97	99.40	28	27.88	99.57
5	5	4.86	97.20	35	34.85	99.57
6	10	9.97	99.70	7	6.5	92.86
7	10	9.86	98.60	14	13.89	99.21
8	10	9.88	98.80	21	19.95	95.00
9	10	9.89	98.90	28	27.98	99.93
10	10	9.85	98.50	35	34.89	99.69
11	15	14.96	99.73	7	6.95	99.29
12	15	14.97	99.80	14	13.98	99.86
13	15	14.86	99.07	21	20.95	99.76
14	15	14.92	99.47	28	27.98	99.93
15	15	14.96	99.73	35	34.96	99.89
16	20	19.86	99.30	7	6.96	99.43
17	20	19.87	99.35	14	13.97	99.79
18	20	19.99	99.95	21	20.97	99.86
19	20	19.95	99.75	28	27.96	99.86
20	20	19.96	99.80	35	34.98	99.94
21	25	24.96	99.84	7	6.85	97.86
22	25	24.86	99.44	21	20.91	99.57
23	25	24.92	99.68	28	27.94	99.79
24	25	24.86	99.44 Mean =99.21 SS =0.720	35	34.85	99.57 Mean =99.43 SS =0.68

This study, the statistical parameters were found to produce a satisfactory validity for the PLS and PCR methods. The PLS and PCR methods have reliable accuracy and higher precision.

For calibration the prediction residual error sum-ofsquares (PRESS) was calculated as:

$$PRESS = \sum_{i=1}^{n} (C_i^{added} - C_i^{found})^2$$
 (3)

 C_i^{added} : True Concentration, additional concentration of drug substance C_i^{added} : Estimated

Concentration, calculate concentration of drug substance

Some statistical parameters determined the effectiveness of the calibration. The standard error of prediction (SEP) was calculated using the following expression:

$$SEP = \sqrt{\frac{\sum_{i=1}^{n} (C_i^{added} - C_i^{found})^2}{n-1}}$$
(4)

 C_i^{added} : Actual Concentration, the added concentration of drug, C_i^{added} : Predicted

Concentration, the calculated concentration of drug, n: the total number of synthetic mixtures

According to the actual and predicted concentrations of the samples, SEP and PRESS values of metronidazole and moxifloxacin were calculated and listed in Table 3.

Analysis of real samples: Table 4 lists the experimental results of the two numerical methods for commercial products and as you can see the obtained results are very close to each other (Avelox and Flagyl).

Table 3: Statistical parameter values for calibration step- simultaneous determination of metronidazole and moxifloxacin using partial least square and principal component regression methods.

PARAMETER	METHOT	METRONIDAZOLE	MOXIFLOXACIN
SEP	PLS	0.036	0.025
	PCR	0.043	0.028
PRESS	PLS	0.052	0.0096
	PCR	0.064	0.0096

Table 4: Determination of metronidazole and moxifloxacin in commercial products using PLS and PCR methods.

	METRONIDAZOLE(GRAM)		MOXIFLOXACIN(GRAM)		
NO	PLS	PCR	PLS	PCR	
1	0,495	0,496	0,435	0,436	
2	0,49	0,492	0,436	0,434	
3	0,496	0,497	0,43	0,432	
4	0,497	0,493	0,432	0,435	
5	0,496	0,495	0,434	0,435	
Mean	0,4948	0,4946	0,4334	0,4344	
Standard					
Deviation	0,002775	0,002074	0,002408	0,001517	

In this study, chemometric methods can be applied to simultaneously measure spectra data processing based metronidazole and moxifloxacin samples containing mixtures of homogeneously mixed binary drug samples.

In order to compare the performances of the investigated chemometric techniques according to UV spectrophotometric method for real samples we applied Snedecor's *F*-test.

The method used to compare the differences between the one-way ANOVA tests was applied to real samples of the drug sample. In this study, F values of Snedecor were calculated and compared with F values (p = 0.05). The ANOVA test results for metronidazole were 0.0014 (PLS) and 0.0020 (PCR). The ANOVA test results for moxifloxacin 0.0013 (PLS) and 0.0017 (PCR). Experimental (calculated) F values did not exceed F-value (4.05) in analysis of variance. This is the result of a meaningful difference between all these methods. All statistical parameters and numerical values are suitable for real-time simultaneous identification.

4. Conclusions

The partial least squares method and the main component regression, which were successfully applied, were able to identify the active ingredients in the synthetic solutions separately. For all values, low prediction errors and high correlation coefficients emphasize the high linear relationship between predicted and actual concentrations. The results obtained with this binary mixture and some ratios of component concentrations indicate excellent prediction ability with these methods.

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